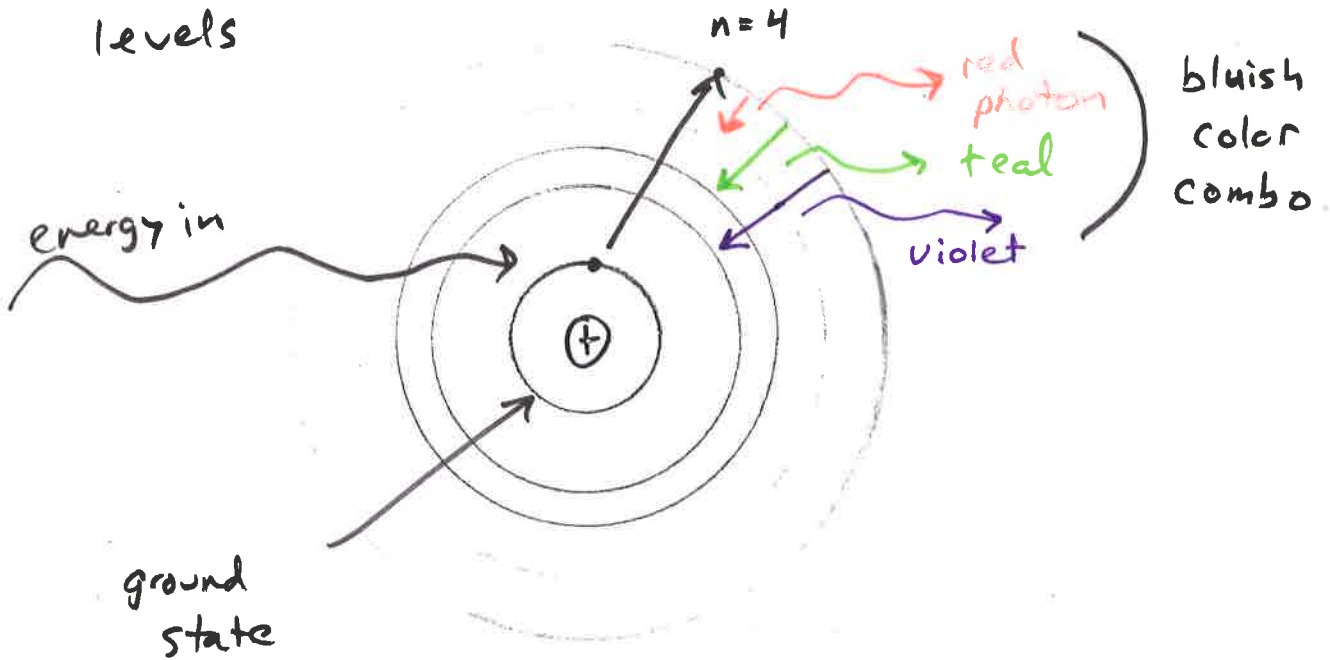


11/16/15 Notes: Niels Bohr Model (1913)

- atom has electron energy levels set at certain distances/energies from the nucleus
- when excited, electron absorbs energy and jumps away from nucleus
- $e^-$  must absorb an energy amt that is the same as a distance between levels



- each element has its own energy level scheme, releasing different energies/colors

